# ISOVALERIC ACID



#### 1 PERFORMANCE

1) Measuring range : 3-50 ppm Number of pump strokes: 1(100mL)

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit

: 3 years 4) Shelf life 5) Operating temperature :  $15 \sim 25$  °C

6) Reading : Graduations printed on the tube are calibrated by Acetic acid

at 1 pump stroke and Isovaleric Acid concentration is determined by

using a conversion chart at 1 pump stroke

7) Colour change : Pale pink  $\rightarrow$  Yellow

## 2. RELATIVE STANDARD DEVIATION

RSD-low 10% RSD-mid 10% RSD-high 10%

#### 3 CHEMICAL REACTION

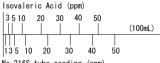
By reacting with alkali, PH indicator is discoloured.  $(CH_3)_2 CHCH_2 COOH + NaOH \rightarrow (CH_3)_2 CHCH_2 COONa + H_2O$ 

# 4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

## 5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxid	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"



No. 216S tube reading (ppm)