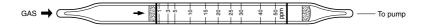
# MONOETHANOL AMINE



### 1. PERFORMANCE

1) Measuring range Number of pump strokes 1  $(100 \text{m} \ell)$  2  $(200 \text{m} \ell)$  2  $(200 \text{m} \ell)$  3) Detectable limit 2  $(200 \text{m} \ell)$  2  $(200 \text{m} \ell)$  4) Shelf life 2 years

5) Operating temperature 2 years  $0 \sim 40 \text{ °C}$ 

6) Temperature compensation : Necessary (See "TEMPERATURE CORRECTION TABLE") 7) Reading : Direct reading from the scale calibrated by 1 pump stroke

8) Colour change : Pink→Pale purple

#### 2. RELATIVE STANDARD DEVIATION

#### 3. CHEMICAL REACTION

By reacting with Sulphuric acid, PH indicator is discoloured. H2NCH2CH2OH + H2SO4→ (HOC2N5NH3) 2SO4

### 4. CALIBRATION OF THE TUBE

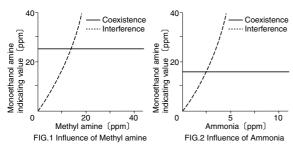
COLOURIMETRY METHOD

#### 5. INTERFERENCE AND CROSS SENSITIVITY

Substance		Interference	Coexistence		
Other amines	FIG.1	Brownish yellow stain is produced.	Double-stain layer (Brownish yellow and Pale purple) is produced.		
ammonia	FIG.2	Similar stain is produced.	Double-stain layer (Yellow and Pale purple) is produced.		
Hydrazine		"	"		
Atmospheric air (CO <sub>2</sub> +H <sub>2</sub> O)		//			

#### (NOTE)

- In case of 2 pump strokes, following formula is available for the actual concentration.
   Actual concentration = 1/2 × Temperature corrected value
- In any case, Pale purple stain only is indicated concentration of Monoethanal amine. Other discolourations
  may be disregarded.



## TEMPERATURE CORRECTION TABLE

Scale	True Concentration (ppm)					
Readings	0℃	10 °C	20℃	30 °C	40°C	
(ppm)	(32°F)	(50°F)	(68° F)	(86° F)	(104° F)	
50	_	-	50	35	29	
40	_	65	40	30	25	
30	-	49	30	23	20	
25	_	39	25	20	17	
20	65	30	20	16	14	
15	45	22	15	12	10	
10	29	14	10	8	7	
5	12	7	5	4	3	
3	6	4	3	3	2	
1	1	1	1	1	1	